

L-Proline-catalyzed intramolecular cyclization of 5-hydroxypentene to β -halogenated tetrahydrofuran

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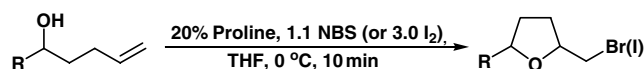
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Abstract—A series of β -bromo- and β -iodotetrahydrofurans was synthesized from the reaction mixture of 5-hydroxypentene, L-proline, NBS (or I₂) in THF at 0 °C for 10 min. This L-proline-catalyzed intramolecular cyclization provides a simple and efficient method for the preparation of β -halogenated tetrahydrofuran.

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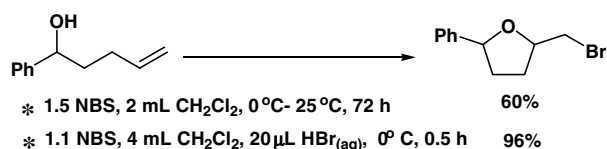
The tetrahydrofuran ring is a very important structural moiety, which is present in a large variety of natural products such as polyether antibiotics.^{1–4} The intramolecular alkoxylation reaction, the addition of alcohol onto carbo-carbon multiple bonds, provides an efficient and direct access to tetrahydrofuran functionality.⁵ 5-Hydroxypentenes have received much more attention as synthetic intermediates for synthesis of this class of heterocycles.^{6–16} The intramolecular cyclization (5-*exo*-trig) of 5-hydroxypentene to tetrahydrofuran ring has been reported and achieved by activation of alkene from halogen^{17–20} or Lewis acid,^{21,22} epoxidation of alkene followed by intramolecular C–O bond formation,^{23–26} and palladium-catalyzed cyclization.^{27–29} Halogenated tetrahydrofurans, especially derivatives with an exocyclic bromo- or iodo- functionality located in the β -position to the ring oxygen atom, have become attractive synthesis targets because of the discovery of β -brominated tetrahydrofurans, which were occurred widely as secondary metabolites in the marine environment.³⁰ The most simple and direct procedure for the synthesis of β -halogenated tetrahydrofurans is the reaction of 5-hydroxypentene with molecular halogen (Br₂, I₂) or halogenating reagent (NBS, IBr, ICl) via intramolecular cyclization (5-*exo*-trig). Herewith, we wish to report a simple and highly efficient method for synthesis of β -bromo and β -iodotetrahydrofuranyl compounds from 5-hydroxypentenes (Scheme 1).



Scheme 1.

5-Hydroxypentenes were prepared by the reaction of 4-bromobutene with aldehydes under a sonochemical Barbier reaction condition.³¹ Thus, we firstly investigated the bromocyclization of 1-phenylpentenol with NBS (*N*-bromosuccinimide)²⁰ at room temperature and β -bromo-phenyltetrahydrofuran was produced in 60% yield after 3 days. The small amount of HBr added to NBS liberates more Br₂ molecule³² and this reaction condition was investigated for bromocyclization of 1-phenylpentenol. The trace amount of concd HBr (20 μ L/equiv) was added to a reaction mixture of 1-phenylpentenol and NBS in CH₂Cl₂ at 0 °C and 96% yield of β -bromo-phenyltetrahydrofuran was obtained after 30 min (Scheme 2). The catalytic amount of HBr promoted the formation yield of β -bromotetrahydrofuran and the reaction time decreased dramatically.

A series of 5-hydroxypentenes was investigated under this HBr-catalyzed bromocyclization reaction condition



Scheme 2.

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Table 1. HBr-catalyzed bromocyclization of 5-hydroxypentene

Entry	R	Product	Yield ^a (cis/trans) ^b
1			72% (26:74)
2			88% (44:56)
3			92% (26:74)
4			96% (30:70)
5			95% (32:68)
6			99% (35:65)
7			98% (39:61)
8			75% (40:60)
9			64% (30:70)
10			93% (37:63)
11			99% (37:63)
12			80% (42:58)

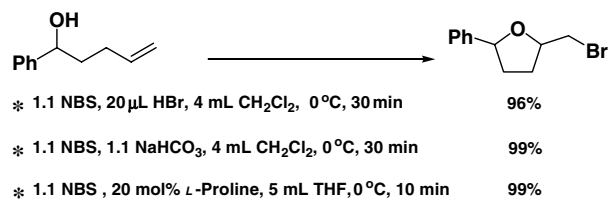
^a The yields were determined after chromatographic purifications.

^b The cis/trans ratio is determined by ¹H NMR spectral analysis.

and the results are shown in Table 1. The typical procedure for synthesis of a β-bromotetrahydrofuran is as follows: To a reaction mixture of 5-hydroxypentene (1.0 mmol) and NBS (1.1 mmol) in CH₂Cl₂ at 0 °C was added concentrated HBr (20 μL). After the reaction mixture was stirred at 0 °C for 30 min, water (10 mL) was added and extracted with ether (3 × 20 mL). The combined organic layer was washed with Brine (30 mL), dried with MgSO₄, filtered, and then the organic solvent was removed under reduced pressure. Further purification was achieved on a flash chromatograph with silica gel and ethyl acetate/hexanes.

Good to excellent yields of the investigated compounds in Table 1 were afforded under this HBr-catalyzed reaction condition. The diastereoselectivity (cis/trans) ratios were also determined by ¹H NMR spectral analysis. The coupling constant values of 2- and 5-position protons were measured and as reference for the determination of cis/trans ratio.

The addition of HBr catalyzed the bromocyclization of 5-hydroxypentene. Thus, we think the addition of a base may accelerate the cyclization step by deprotonation of alcohol which becomes a better nucleophile to the acti-

**Scheme 3.**

vated carbon–carbon double bond. The stoichiometric amount of a base such as NaHCO₃ was added to a mixture of 5-hydroxypentene and NBS and it was stirred at 0 °C for 30 min (Scheme 3). The introduction of a base improves the reaction rate and yield which presents the same effect as the addition of HBr. Thus, we think the introduction of an amino acid such as proline^{33–36} may exhibit the similar improvement for this intramolecular cyclization. To a reaction mixture of 1-phenylpentenol and NBS in THF at 0 °C was added natural proline. The reaction mixture was stirred at 0 °C for 10 min and 99% yield of β-bromo-phenyltetrahydrofuran was obtained. A catalytic amount of L-proline accelerated this bromocyclization of 5-hydroxypentene compound and a high formation yield was also achieved.

A series of 5-hydroxypentenes was investigated under this proline-catalyzed bromocyclization and the results are shown in Table 2. The typical procedure for synthesis of a β-bromotetrahydrofuran in the presence of proline as catalyst is as follows: To a reaction mixture of 5-hydroxypentene (1.0 mmol) and NBS (1.1 mmol) in THF at 0 °C was added L-proline (100 μL, 2 M aqueous solution). After the reaction mixture was stirred at 0 °C for 10 min, water (10 mL) was added, and extracted with ether (3 × 20 mL). The combined organic layer was washed with Brine (30 mL), dried with MgSO₄, filtered, and then the organic solvent was removed under reduced pressure. Further purification was achieved on a flash chromatograph with silica gel and ethyl acetate/hexanes.

The yields and stereoselectivities (cis/trans) of β-bromotetrahydrofuranyl compounds, which were obtained from the L-proline-catalyzed reaction conditions, usually are better than the product obtained by HBr-catalyzed cyclization reaction condition. Alkyl-substituted 5-hydroxypentenes afford much higher yields of tetrahydrofuranyl compounds under proline-catalyzed cyclization reaction condition. Heterocyclic tetrahydrofuranyl compounds were easily synthesized under the reaction conditions.

β-Iodotetrahydrofuranyl compounds were typically prepared from the intramolecular cyclization of 5-hydroxypentenes by using I₂ as iodinating reagent. Thus, we investigated the intramolecular cyclization of 1-phenylpentenol with iodine and the expected product was not produced after 24 h (Scheme 4). Another iodinating reagent IBr was also investigated and generated 68% yield of β-iodotetrahydrofuran after 2 h stirring at 0 °C. The β-iodophenyltetrahydrofuran was obtained with a 54%

Table 2. Proline-catalyzed bromocyclization of 5-hydroxypentene

Entry	R	Product	Yield ^a (cis/trans) ^b
1			98% (24:76)
2			94% (30:70)
3			94% (6:94)
4			99% (35:65)
5			99% (28:72)
6			95% (27:73)
7			98% (24:76)
8			94% (30:70)
9			57% (28:72)
10			86% (26:74)
11			99% (27:73)
12			95% (31:69)

^a The yields were determined after chromatographic purifications.

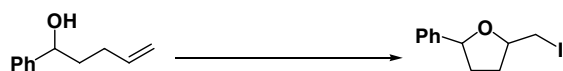
^b The cis/trans ratio is determined by ¹H NMR analysis.

Table 3. Proline-catalyzed iodocyclization of 5-hydroxypentene

Entry	R	Product	Yield ^a (cis/trans) ^b
1			97% (30:70)
2			88% (34:66)
3			96% (25:75)
4			99% (30:70)
5			93% (23:77)
6			99% (24:76)
7			92% (51:49)
8			76% (36:64)
9			61% (39:61)
10			88% (45:55)
11			98% (22:78)
12			90% (44:56)

^a The yields were determined after chromatographic purifications.

^b The cis/trans ratio is determined by ¹H NMR spectral analysis.



* 1.1 I ₂ , 2 mL CH ₂ Cl ₂ , 0 °C, 24 h	N. R.
* 1.1 IBr, 5 mL CH ₂ Cl ₂ , 0 °C, 2 h	68%
* 2.0 I ₂ , 1.0 KI, 5 mL THF, 0 °C, 24 h	54% + S.M.
* 3.0 I ₂ , 1.5 KI, 5 mL THF, 0 °C, 3.5 h	91%
* 3.0 I ₂ , 20 mol% L-Proline, 5 mL THF, 0 °C, 10 min	99%

Scheme 4.

when a base KI was introduced into the reaction mixture. The yield can be increased to 91% when an amount of I₂ and KI were introduced to 3 and 1.5 equiv. An excellent yield (99%) was achieved when a catalytic amount of L-proline was introduced to the iodocyclization reaction condition.

A series of 5-hydroxypentenes was investigated under this proline-catalyzed iodocyclization reaction condition and the results are shown in Table 3. The typical procedure for synthesis of a β-iodonated tetrahydrofuran is as follows: To a reaction mixture of 5-hydroxypentene

(1.0 mmol) and I₂ (3.0 mmol) in THF at 0 °C was added L-proline (0.2 mmol) and the reaction mixture was stirred at 0 °C for 10 min. The mixture was quenched with saturated Na₂S₂O₇ (10 mL), extracted with ether (20 mL × 3). The organic layer was washed with brine (30 mL), dried with MgSO₄, and then removed under reduced pressure. Further purification was achieved on a flash chromatograph with silica gel and ethyl acetate/hexanes.

The experimental results showed that the formation yield of β-iodotetrahydrofuran was nearly as high as the yield of β-bromotetrahydrofuran obtained under the reaction condition. The stereoselectivity (cis/trans) of proline-catalyzed bromocyclization reaction generally is better than its corresponding iodocyclization reaction.

In conclusion, this natural proline-catalyzed reaction provides a simple and highly efficient method for the preparation of β-halogenated (β-bromo or β-iodo) tetrahydrofuranyl compound from 5-hydroxypentene via an intramolecular cyclization (5-exo-trig) under either acidic or basic reaction condition.

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